

## Ground Water Quality Assessment of Marble Mining Areas in Rajsamand District, Rajasthan, India

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### Abstract

The present study represents the impact of marble mining and processing units on the quality of ground water in the Rajsamand district of Rajasthan state. For this work various water samples were collected from surrounding areas of the mining hub covering all the tehsil namely- Rajsamand, Amet, Bhim, Deogarh, Khamnor, Kumbhalgarh, and Railmangra of Rajsamand district. The sample were analysed for various Physio- chemical parameters like-Electrical Conductivity(EC), pH, Total Hardness (TH), Dissolved Solids-Total (TDS), Sodium ( $\text{Na}^{+1}$ ) Potassium( $\text{K}^{+1}$ ), Calcium ( $\text{Ca}^{+2}$ ), Magnesium ( $\text{Mg}^{+2}$ ), Chloride ( $\text{Cl}^{-1}$ ), Sulphate ( $\text{SO}_4^{-2}$ ), Carbonate ( $\text{CO}_3^{-2}$ ), Bicarbonate ( $\text{HCO}_3^{-2}$ ), Nitrate( $\text{NO}_3^{-1}$ ), Fluoride (F), the result of water quality parameters was compared with IS:10500-2012 drinking water specification. The finding of results indicates that the level of the certain parameters like TDS, TH, Ca, Mg, Na, K, Cl,  $\text{NO}_3$ , and F exceed the limits of ground water stipulated by Bureau of Indian Standards (BIS). As per the observations, we must say that it may be possible that one of the egregious sources to polluting ground water in the region is marble mining and its allied activities. The present study is based on the sample collected and tested in the laboratory and it is an attempt to determine the physio- chemical characteristics of ground water in the marble mining area of the Rajsamand district in Rajasthan state of India.



### Article History

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Ground Water;  
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Quality.

### Introduction

Excavation of the mineral carried out by the two major ways that is sub surface mining (Open cast) and underground mining method. In both of the methods ground water play an important role because, due to mining geological pattern of the rocks changes that leads to change in hydrology of

the particular region and ultimately it leads to water pollution or change in water quality & quantity.

There is no question mark that ground water is the important source for survival of agriculture, human health, energy as well as for the ecosystem on entire planet.

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The stress on ground water scarcity, decline of water quality is being increased due to overexploitation and mismanagement of the generated waste water and solid waste in industrial area.

Reported that prolonged mining activities have degrade water quality and decrease recharging capacity of the area<sup>1</sup> cited that drinking water must be free from microorganisms and should have appropriate chemical concentrations imbalance in both of the properties cause health impacts.<sup>2</sup>

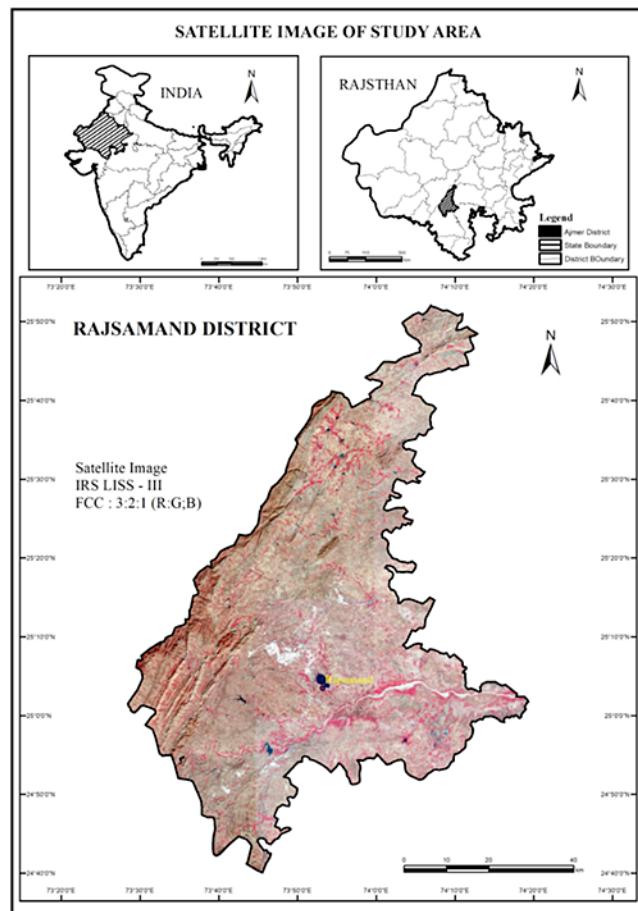
In this article the objective of the study is to identify the changes and extent of contamination in ground water quality in and around the mining activities of the of the study area.

The study area is facing serious threat on ground water quality degradation as the entire district have

ample number of small-scale mining unit running in unsystematic manner. Water samples were collected from nearby areas of mining unit and as per the samples collected and analysed the overall water quality of the entire district is suitable for drinking purpose only after adequate treatment. As per the ground water atlas book 2013 all 7 blocks of Rajsamand district falls in over exploited zone it will cause serious water scarcity in near future.

**Study Area**

The Rajsamand district is situated in southern part of the Rajasthan. It shares the North boundary withAjmer and East with Bhilwara districts, South by Udaipur, in west it is bounded by Pali. It stretches between 23° 31' 49.64 to 24° 30' 16.57 North latitude and 74° 13' 19.93 to 74° 58' 59.58 East Longitude the general elevation between 500 to 625 m above mean sea level.



**Fig.1: Location Map of the Rajsamand District**

**Material and Method**

Samples for ground water were collected from open well, tube well, hand pump for physio-chemical analysis during post monsoon-2019. During sampling of the representative water samples IS:3025 standard method was followed and collected samples were transported to the GMEC International laboratory and analysed as per standard methods for various physio-chemical parameters like- Electrical Conductivity, pH, Total Alkalinity, Total Dissolved Solids, Sodium, Potassium, Calcium, Magnesium, Chloride, Sulphate, Carbonate, Bicarbonate, Nitrate, Fluoride, Total Hardness.

**Result and Discussion**

Total 70 Samples were Collected from different blocks (Amet, Bhim, Deogarh, Khamnor, Kumbhalgarh, Railmagra, Rajsamand) for the analysis of basic parameters like Electrical Conductivity, pH, Total Alkalinity, Total Dissolved Solids, Sodium, Potassium, Calcium, Magnesium, Chloride, Sulphate, Carbonate, Bicarbonate, Nitrate, Fluoride, and Total Hardness. Result given in table no 1, all the parameters are manifested in mg/l, pH has no Unit and electrical conductivity (EC) given in Micro Siemens per centimetre

**Table: 1 Physico-Chemical Characterization of Ground Water**

Parameters	Unit	Amet			Bhim			Deogarh		
		Maxim-al	Minim-al	Mean	Maxima-l	Minima-l	Mean	Maxim-al	Minima-l	Mean
Conductivity	µS/cm	580	4100	2340	500	9600	5050	360	3350	1855
pH	N/A	7.6	8.6	8.1	7.9	8.9	8.4	7.9	9	8.45
TDS	mg/l	313	2300	1307	286	5463	2875	179	1896	1038
TH	mg/l	130	1195	663	130	2260	1195	95	625	360
Na <sup>+1</sup>	mg/l	49	469	259	43	1139	591	32	467	249.5
K <sup>+1</sup>	mg/l	4	196	100	7	74	40.5	4	383	193.5
Ca <sup>+2</sup>	mg/l	22	178	100	36	381	209	16	132	74
Mg <sup>+2</sup>	mg/l	11	182	97	10	319	165	13	72	43
Cl <sup>-1</sup>	mg/l	89	979	534	53	3085	1569	14	723	369
SO <sub>4</sub> <sup>-2</sup>	mg/l	10	432	221	19	394	207	5	303	154
CO <sub>3</sub> <sup>-2</sup>	mg/l	0	30	15	0	99	49.5	0	120	60
HCO <sub>3</sub> <sup>-1</sup>	mg/l	183	464	324	73	415	244	159	543	351
NO <sub>3</sub> <sup>-1</sup>	mg/l	0	129	64.5	10	112	61	6	175	90.5
F	mg/l	0.82	2.98	1.9	0.42	3.94	2.18	0.2	5.65	2.93

Parameters	Unit	Khamnor			Kumbhalgarh			Railmagra		
		Maxim-al	Minim-al	Mean	Maxima-l	Minima-l	Mean	Maxim-al	Minima-l	Mean
EC	µS/cm	520	7080	3800	620	5650	3135	1080	8000	4540
pH	N/A	7.6	8.4	8	7.8	8.4	8.1	7.7	9	8.35
TDS	mg/l	275	3946	2111	317	3084	1701	555	4845	2700
TH	mg/l	120	1905	1013	70	1580	825	185	1340	763
Na <sup>+1</sup>	mg/l	58	731	394.5	40	559	299.5	90	1316	703
K <sup>+1</sup>	mg/l	0	121	60.5	4	31	17.5	5	90	47.5
Ca <sup>+2</sup>	mg/l	16	405	211	6	136	71	20	224	122
Mg <sup>+2</sup>	mg/l	19	218	119	13	302	158	27	207	117
Cl <sup>-1</sup>	mg/l	78	2184	1131	64	1652	858	71	1744	908
SO <sub>4</sub> <sup>-2</sup>	mg/l	10	298	154	0	250	125	48	1249	649
CO <sub>3</sub> <sup>-2</sup>	mg/l	0	15	7.5	0	24	12	0	78	39
HCO <sub>3</sub> <sup>-1</sup>	mg/l	116	384	250	134	391	263	177	586	382

NO <sub>3</sub> <sup>-1</sup>	mg/l	4	149	76.5	0	63	31.5	25	175	99.5
F	mg/l	0.36	1.52	0.94	0.3	4.61	2.46	0.67	3.76	2.22

Parameters	Unit	Rajsamand		
		Minimal	Maximal	Mean
EC	µS/cm	1470	7000	4235
pH	N/A	7.6	8.8	8.2
TDS	mg/l	805	4083	2444
TH	mg/l	225	1550	888
Na <sup>+1</sup>	mg/l	69	888	478.5
K <sup>+1</sup>	mg/l	6	360	183
Ca <sup>+2</sup>	mg/l	18	220	119
Mg <sup>+2</sup>	mg/l	44	243	144
Cl <sup>-1</sup>	mg/l	170	1773	972
SO <sub>4</sub> <sup>-2</sup>	mg/l	62	1239	651
CO <sub>3</sub> <sup>-2</sup>	mg/l	0	120	60
HCO <sub>3</sub> <sup>-1</sup>	mg/l	153	683	418
NO <sub>3</sub> <sup>-1</sup>	mg/l	0	218	109
F	mg/l	0.06	2.61	1.34

**Table: 2 Comparison of Ground Water Quality Data with Drinking Water Standard**

Sr No	water Quality parameters and unit	WHO International Standards 2017 guidelines		BIS-10500:2012		Sample Range in Study Area
		Desirable Limit	Maximum Permissible Limit	Desirable Limit	Maximum Permissible Limit	
1.	EC (µS/mg)	1400	-	-	-	360-9600
2.	pH	6.5	8.5	6.5	8.5	07.6- 9.0
3.	TDS (mg/l)	500	1000	500	2000	179 -5463
4.	TH (mg/l)	100	500	200	600	70-2260
5.	Na <sup>+1</sup> (mg/l)	-	200	-	-	32-1316
6.	K <sup>+1</sup> (mg/l)	-	12	-	-	00-383
7.	Ca <sup>+2</sup> (mg/l)	75	200	75	200	6.0 - 405
8.	Mg <sup>+2</sup> (mg/l)	50	150	30	100	10.0 -319
9.	Cl <sup>-1</sup> (mg/l)	200	600	250	100	14-3085
10.	SO <sub>4</sub> <sup>-2</sup> (mg/l)	200	400	200	400	62-1249
11.	CO <sub>3</sub> <sup>-2</sup> (mg/l)	-	-	-	-	00-120
12.	HCO <sub>3</sub> <sup>-1</sup> (mg/l)	-	-	-	-	73-683
13.	NO <sub>3</sub> <sup>-1</sup> (mg/l)	-	50	45	No Relaxations	00-218
14.	F (mg/l)	0.5	1	1	1,5	0.82-5.65

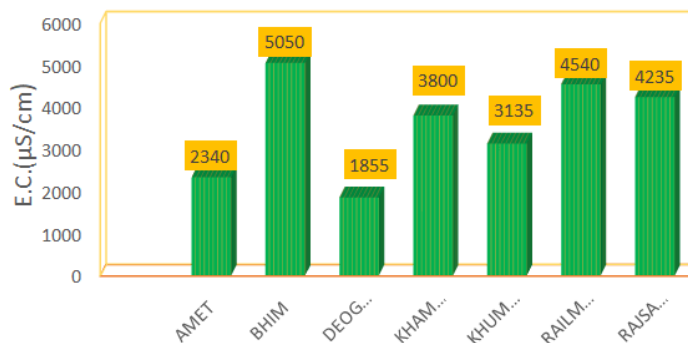
The results were compared with drinking water standards for the parameters like - Conductivity, pH, Dissolved Solids-Total, Total Alkalinity Sodium,

Potassium, Calcium, Magnesium, Chloride, Sulphate, Carbonate, Bicarbonate, Nitrate, Fluoride, and Total Hardness describe in table 2.

**Electrical Conductivity**

As per the result obtained value of conductivity ranges minimum- 360  $\mu\text{S}/\text{cm}$  (Anjana village of Deogarh tehsil) maximum -9600  $\mu\text{S}/\text{cm}$  in Jujupura village of Bhimtehsil. Almost 90% of the samples

exceeds the desirable limit of WHO drinking water standards. Higher concentration of acid, base and salts will be the result as higher EC.<sup>3</sup> Higher values of EC also indicate presence of dissolved inorganic substances in ionized form at higher level.<sup>4</sup>

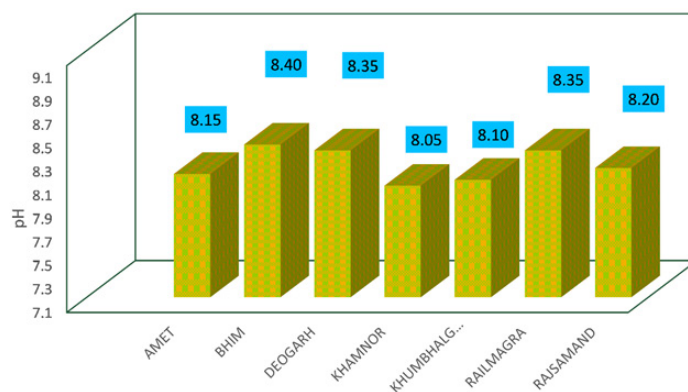


**Fig. 2: Graphical representation of EC (Mean Values) observed in GW of Study Area**

**pH**

As per result of collected samples the pH value ranges between 7.6 pH at Pasaoond Village of Rajsamand tehsil to 9.0 pH at Kuraj Village in Railmagaratehsil standard limit of pH for drinking water is 6.5 to 8.5 (BIS 10500 -2012). 90% of the water samples are well under the prescribed limit of

drinking water specification except 10% of sampling station shows higher pH values. There is no direct impact of higher pH on human health but it regulates all the biological activities in the water however “higher pH values reduce the disinfection potential of the chlorine.”<sup>5</sup>



**Fig. 3: Graphical representation of pH (Mean Values) observed in GW of Study Area**

**Total Dissolved Solid (TDS)**

As shown in table-1 values of the TDS ranges between 179 mg/l at Anjana village sampling station to 5463 mg/l at Jujupura sampling station with the average value of 1410 mg/l, more than 90% of the sample results found beyond the desirable limit only few sampling station were found well within the desirable limit 6 reported that higher values of

TDS in ground water is due to leaching of salt from soil. As we earlier discussed unsystematic dumping cause leaching of mineral in ground water which leads to increase in dissolved solid in ground water of the nearby vicinity. Higher values of TDS (more than 1500 PPM) cause gastrointestinal irritation.<sup>7</sup> Earlier reported that the mean amount of TDS in groundwater samples that have been exposed

to marble industrial wastes was much higher in comparison to control sample group and results compared with previous studies and found that the

ground water is contaminated by marble industry waste products.<sup>8</sup>

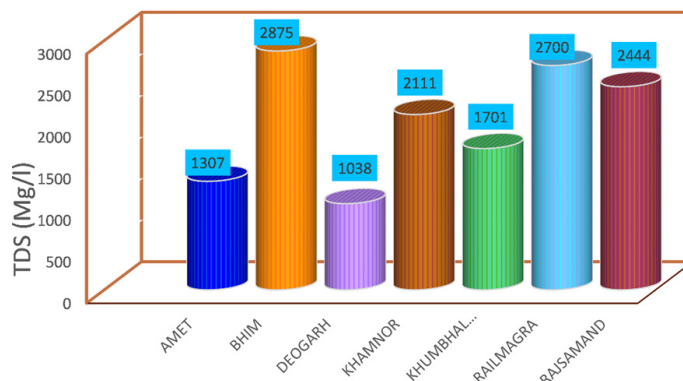


Fig. 4: Graphical representation of TDS (Mean Values) observed in GW of Study Area

**Total Hardness**

From the analysis results it is found that total hardness ranges between -70 mg/l to 1905 mg/l, only 18% of the sample are below the desirable limit of BIS drinking water specification rest 78% of the samples were beyond the desirable limit that is 300 to 600 mg/l for drinking water and 70% water samples are below the permissible limit at

boundary line. Higher values of soluble salt of Ca, Mg, carbonate and bicarbonate ions along with chloride and sulphate governs the water quality and make it unsuitable for drinking as well as potable use due to hard nature it may be temporary hardness or permanent hardness it depends on bonding of anions with Ca and Mg the ill impact of total hardness is stone formation in kidney and heart diseases.<sup>9</sup>

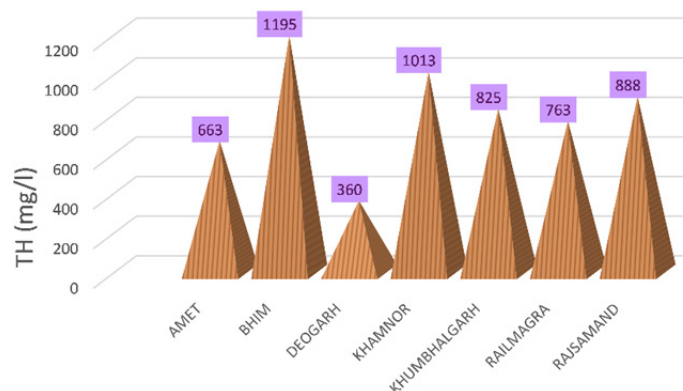


Fig. 5: Graphical representation of TH (Mean Values) observed in GW of study area

**Sodium (Na<sup>+</sup>)**

As we know in the order of element abundance Sodium is on sixth number and generally it found soluble form in natural water. Natural source of sodium in water is rocks and soils, Large amounts of sodium in combination with chloride give a salty taste. High concentration of sodium in water makes unsuitable for irrigation and sodium salts

also responsible for foaming in steam boilers.<sup>10</sup> The result obtained in the study area as describe in table 1 concentration of sodium in ground water samples found in the range of 32 mg/l at Anjana Village of Deogarh tehsil to 1139 mg/ l at Jujupra village of Bhim tehsil. Maximum permissible standard for sodium is prescribed 200 mg/l by WHO in drinking water. Result obtained from the analysed

sample on an average 60% of the water samples found above the prescribed permissible standard. Water intake with high concentration of sodium may

cause high blood pressure arteriosclerosis, oedema and hyperosmolarity.<sup>6</sup>

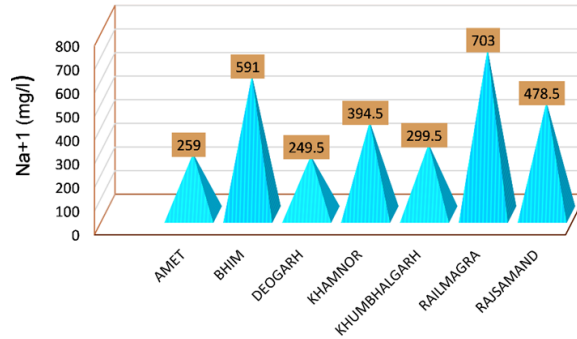


Fig. 6: Graphical representation of Sodium (Mean Values) observed in GW of study area

**Potassium (K<sup>+</sup>)**

According to present study concentration of potassium found lower than sodium except few locations result ranges from 0 mg/l at Nathdwara village in Khamnor tehsil to 383 mg/l at Tal village in

Deogarh tehsil. As per WHO maximum permissible standard limit for potassium in drinking water is 12 mg/l all the water sample exceed WHO maximum permissible standard limit in the sampling areas.

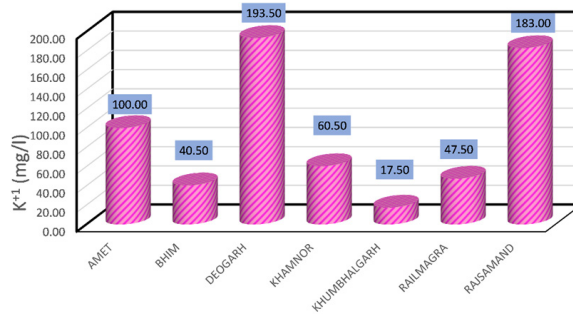


Fig. 7: Graphical representation of Potassium (Mean Values) observed in GW of Study Area

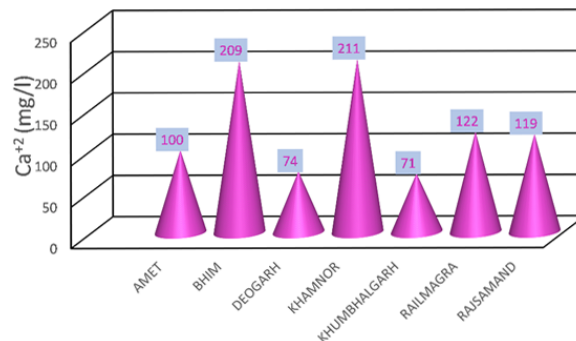


Fig. 8: Graphical representation of Calcium (Mean Values) observed in GW of study area

**Calcium (Ca<sup>2+</sup>)**

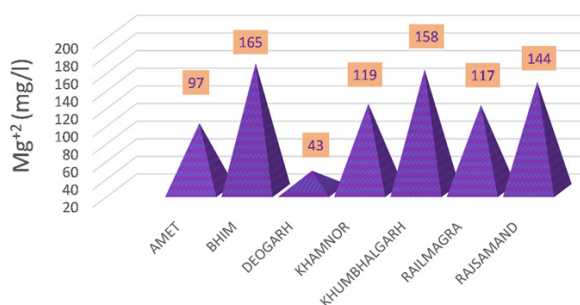
According to present study and result obtained concentration of the calcium found in the range of

6mg/l at Kelwara village to maximum 405 mg/l at Koonchal village among all the sampling station around 60% samples exceed acceptable limit of



BIS drinking water standard that is 75 mg/l<sup>7</sup> only 40 % sample found below the acceptance limit and 0 .5 % sample found beyond the maximum permissible standard limit of BIS potable water standard that is 200 mg/l<sup>7</sup> consumption of higher concentration of calcium for prolong period lead

to abdominal illness and make it unfit for domestic purpose due to its encrustations and scaling nature. In the sampling area higher concentration of Ca<sup>+2</sup> is result of the leaching of mineral from marble mining and dumping areas in the ground water.

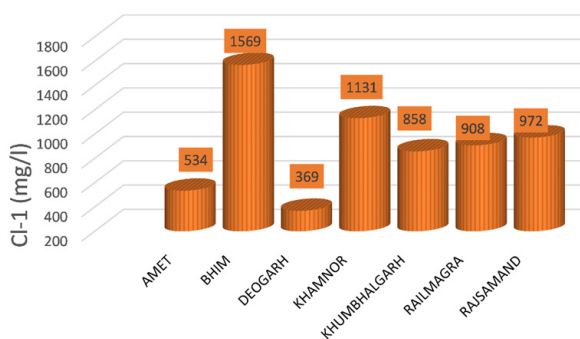


**Fig. 9: Graphical representation of Magnesium (Mean Values) observed in GW of study area**

**Magnesium (Mg<sup>+2</sup>)**

Magnesium is the most abundant mineral along with calcium, according to result obtained it found in the range of 10 mg/l at Padmela village to 319 mg/l at Jujupura village in the present study 90% of the sample found beyond the desirable limit

of BIS drinking water standard that is 30 mg/l<sup>7</sup> and 85 % of the water samples found below the maximum permissible standard limit of BIS drinking water standard which is 100 mg/l<sup>7</sup> higher concentration of the Magnesium is undesirable for domestic purpose as it is responsible for harness of the water.



**Fig. 10: Graphical representation of Chloride (Mean Values) observed in GW of study area**

**Chloride (Cl<sup>-1</sup>)**

Chloride plays a vital role to identifying water quality, source of chloride in ground water may be weathering of rock, sedimentary rock leaching, domestic and industrial waste as well as fertilizers in agriculture field. In the present study chloride found in the range of 14 mg/l at Anjna village to 3085 mg/l at Jujupura village however mean value found 784 mg/l for all stations. Desirable limit for chloride is 200mg/l as per drinking water standard given in 2012 by Bureau of Indian standard and maximum permissible limit is 600 mg/l the study conducted shows only

37 % samples are under the presumable limit around 1-2 % samples found above the maximum presumable limit of BIS standard.

**Sulphate (SO<sub>4</sub><sup>-2</sup>)**

Sulphate Commonly present in mine water and in some industrial wastes, higher concentration of Sulphate has a laxative effect and combination with other ions, give a bitter taste.<sup>10</sup> Sulphate found in natural water in anion form as per BIS: 10500, 2012 presumable limits for sulphate in potable water is 200 mg/l and 400mg/l is the maximum



presumable standard in the absence of alternative source. Sulphate is measure by turbidity method by using nephelometer or spectrophotometer meter by using barium chloride and conditioning reagent. In the present study sulphate found between 0 mg/l to 1239 mg/l, among all the samples 75 % samples found below the permissible limit and 92% samples found

under the limit of maximum presumable standard limit for potable water. At few sampling stations shows high concentration of sulphate which may create scale formation in water pipes and may be gives piquant taste in water that can have a purgative effect on human health as well on livestock

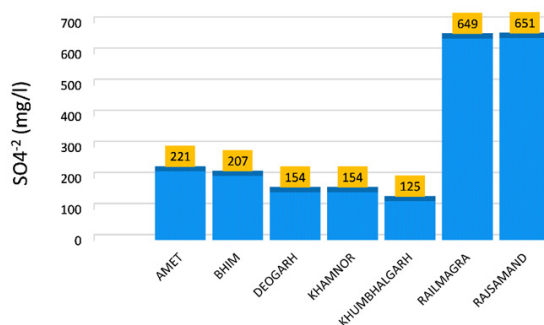


Fig. 11: Graphical representation of Sulphate (Mean Values) observed in GW of Study Area

**Carbonate (CO<sub>3</sub><sup>-2</sup>)**

The value of carbonate is found in the range of 0 to 120 milligram per litre among all the sampling station and mean value found 33 mg/l for all the stations.

Almost 50% samples show nil values of carbonate only two sampling points indicate more than 100 mg/l of carbonate.

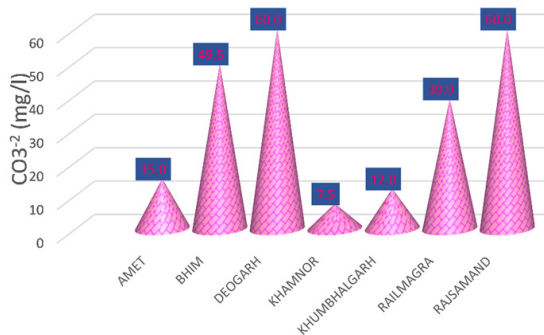


Fig. 12: Graphical representation of Carbonate (Mean Values) observed in GW of Study Area

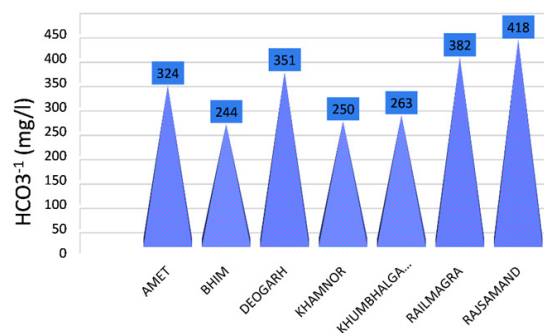
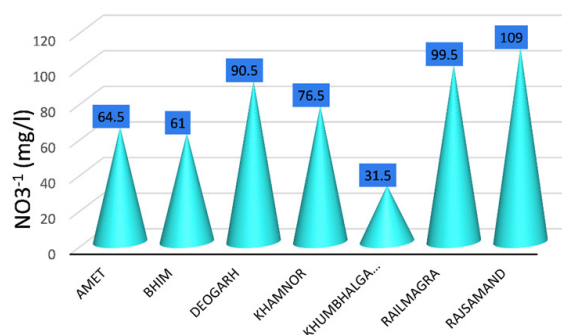


Fig. 13: Graphical representation of Bicarbonate (Mean Values) observed in GW of study area

**Bicarbonate ( $\text{HCO}_3^{-1}$ )**

The present study stated that the concentration of bicarbonate ranges from 73 mg/l at Jujpurato 683 milligram per litre at Bhatoli in Rjsamand Tehsil while

mean value of all the samples was found 122 mg/l. (Stumm and morgan 1996) studies that dominance of mineral dissolution is responsible for higher concentration of bicarbonate in the ground water.



**Fig. 14: Graphical representation of Nitrate (Mean Values) observed in GW of study area**

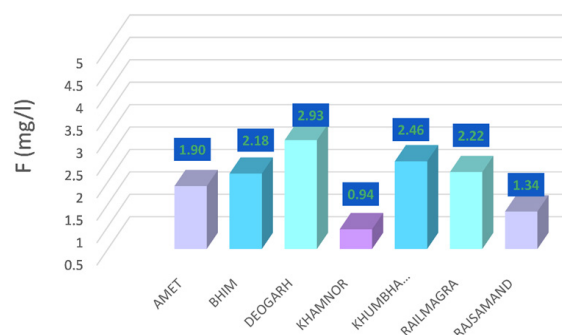
**Nitrate ( $\text{NO}_3^{-1}$ )**

As per the study conducted the range of NO<sub>3</sub> found in the samples from 0 milligram per litre to 175 milligram per litre while permissible limit of nitrate as per BIS is 45 mg/l there is no relaxation in the standard presumable limit above 45 mg/l in potable water. 60% samples found within the permissible limit of drinking water standard and 40% samples exceed the permissible limit during study period it is already reported that higher concentration of the nitrate causes blue baby syndrome in child. Source of nitrate in ground water are fertilizer, domestic sewage and industrial waste.

the presence of concentration of fluoride accessory minerals in the rock mineral assemblage through which the ground water is flow<sup>11</sup> In the present studies fluoride found 0.06 milligram per litre to 5.65 milligram per litre while mean value of all the samples were found 1.93 milligram per litre. as per BIS presumable limit for potable water for fluoride is 1 mg/l and maximum presumable standard limit is 1.5 milligram per litre from the result obtained it is indicated that 62% samples found within the presumable limit while 38% samples exceed the acceptable limit while 90% of the samples found within the maximum presumable limit significant source of fluoride in ground water is leaching, weathering of soluble fluoride. As per earlier studied excess fluoride cause dental, skeletal and non-skeletal fluorosis.

**Fluoride (F)**

In nature fluoride occurs in the form of fluorspar (fluorite), rock phosphate, triphite, phosphorite crystals climate is the major factor which control



**Fig. 15: Graphical representation of Fluoride observed in GW of Study Area**

### Conclusion

Based on Sample collected and analysed it could be summarized that quality of the ground water is deteriorated due to marble mining and its allied activities in entire district, the measured criterion is Electrical Conductivity, pH, Alkalinity-Total, Dissolved Solids-Total, Sodium, Potassium, Magnesium, Calcium, Sulphate, Chloride, Carbonate, Bicarbonate, Nitrate, Fluoride, and Total Hardness. Obtained results indicated that the Contamination in ground water quality is the likely result of unsystematic dumping, of marble and processing waste and irregular and unplanned mining activities. Reported that Concentration of TDS, Hardness, alkalinity, calcium and sulphate including EC and pH has deteriorated due to the mining and its allied activities.<sup>12</sup> Reported that leachates can enhance the dissolution of inorganic materials or substances such as chlorides, bicarbonates, sodium, sulphate and potassium content of water.<sup>13</sup> reported that waste from marble industry destroying the ground water resources in the Zarqa Governorate, Jordan.<sup>14</sup>

The overall water samples can only be suitable for drinking purpose after further necessary treatment. degradation in water quality will pose serious health hazard to the local people.<sup>10</sup>As per the ground water atlas book 2013 all Seven blocks of Rajsmand district falls in over exploited zone it will cause serious water scarcity in near future

As the entire district have ample number of small-scale marble units which causing serious environmental implications therefore it should be taken in note that actions should be taken to improve the water quality and action plan should be implement to improve water quality of the marble mining areas in the Rajsmand District it is suggest that ground water recharging with rain water could minimize the concentration of minerals and leads to raise the ground water level in area.

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### Conflict of Interest

The authors do not have any conflict of interest.

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